



Short communication

Cu doped graphitic C₃N₄ for *p*-nitrophenol reduction and sensing applications

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ABSTRACT

A series of Cu doped graphitic carbon nitride (*g*-C₃N₄) catalysts with different ratios of copper were synthesized by thermal method, using urea and copper sulphate as precursors. XPS (X-ray Photoelectron Spectroscopy) studies indicated that the peak of N1s at 398.9 eV is due to Cu–N bond and a peak at 932.9 eV showed the presence of Cu–N–C bond. Catalytic hydrogenation of *p*-nitrophenol (Para-NP) to *para*-aminophenol (Para-AP) in presence of *g*-C₃N₄ (*g*CN) and copper doped *g*-C₃N₄ (CN-Cu) with sodium borohydride (NaBH₄) was investigated by UV-Visible spectroscopy. The as-synthesized graphitic carbon nitride with 1% copper (CN-1Cu) doped sample exhibits superior catalytic performance and higher stability compared to fine Cu powder in the second cycle. Additionally, electrochemical paracetamol sensing properties of the samples were studied using cyclic voltammetry and chronoamperometry. Consistent with the catalytic performances, CN-1Cu exhibited greater sensitivity towards the detection of paracetamol within the broad range of 38 μM to 3.64 mM and the study was carried out in the physiological pH ~ 7.4.

1. Introduction

In recent years, graphitic carbon nitride (*g*-C₃N₄) nanosheets, a special class of 2D materials with layered organic frameworks of carbon and nitrogen have received growing and remarkable attention due to their exceptional physicochemical and thermodynamic properties [1]. Carbon nitride is a non-toxic, metal-free and inexpensive semiconductor material, identical to graphene in its two-dimensional layered morphology with carbon to nitrogen in the ratio (C:N) of 3:4 [2,3]. In comparison to other semiconductors such as ZnO and TiO₂, *g*-C₃N₄ can be easily synthesized using a variety of methods and has desirable electrical structure and morphology, as well as high thermal stability up to 600 °C.[4] It has gained significant importance in several emerging

applications such as sensors, water splitting, fuel cells, energy storage devices, and transparent conductors due to their unique properties such as greater charge carrier mobility and high mechanical-thermal-chemical stability [3–8].

Owing to their fascinating properties, *g*-C₃N₄ has been widely being investigated for their catalytic properties by several groups. In one of the recent studies, Gholpour et al. described that mesoporous *g*-C₃N₄ catalysts can be used for the conversion of alcohols in presence of CO₂ and O₂ or N₂ [9]. Sarkar et al. reported that Cu doped *g*-C₃N₄ shows excellent electrocatalytic performance towards oxygen reduction reaction in the basic medium [10]. Mohammadi et al. developed copper oxide nanorods decorated with *g*-C₃N₄ nanosheets for the synthesis of primary amide in water [11]. Covalently bonded sulfamic acid graphitic carbon nitride

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